

Experimental Investigation into Electric Vehicle Battery Diversity

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Abstract— This study conducts experimental investigations into the diversity of electric vehicle (EV) batteries, employing a comparative approach. The primary focus is on analyzing the performance of two distinct battery types under varying states of charge (SOC), utilizing the coulomb counting method. Additionally, a generic battery model is employed to facilitate the comparison process. The experimental setup aims to provide valuable insights into the behavior and characteristics of different EV batteries, contributing to the enhancement of battery technologies for future electric vehicle applications.

Keywords— Experimental Investigations, Electric Vehicle Batteries, Battery Diversity, Comparative Analysis, State of Charge (SOC), Coulomb Counting Method, Generic Battery Model.

I. INTRODUCTION

Recently, electric vehicles have gained enormous popularity due to their performance and efficiency. The investment in developing this new technology is justified by the increased awareness of the environmental impacts caused by combustion vehicles, such as greenhouse gas emissions, which have contributed to global warming and the depletion of oil reserves that are not renewable energy sources.

Battery Management Systems (BMS) must be able to provide an accurate real-time estimate for the Battery State of Charge (SOC) and State of Health (SOH). The SOC value is so critical information for the BMS especially if the battery is the only source of power and the system is not connected to the network. Since the SOC is not directly measurable or captured, it causes the main challenge of finding and proposing different estimation methods.

The literature introduces different approaches for estimating the SOC, this review divides the SOC estimation methods into five categories, which are shown in Figure 1.[1].

In conventional methods, SoC estimation occurs indirectly by combining the physical characteristics of the battery, such as voltage, current, resistance, temperature, and impedance [1-14].

Methods Based on Filtering Algorithms [15-63], Methods Based on Nonlinear Observers [64-68], Methods Based on Learning Algorithms [69-95], Hybrid Methods [96-102].

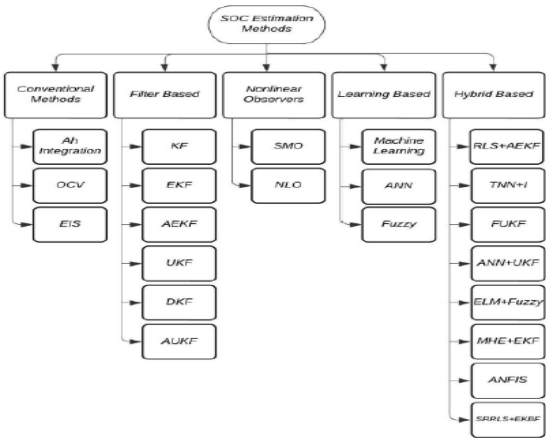


Fig.1. Classification of SOC estimation methods [1]

Table.1 sums up the benefits and drawbacks of four SOC assessment strategies just as their exhibitions in exactness and strength.

TABLE I. COMPARATIVE ANALYSIS OF THE SOC METHODS

Methodology	Method	Main Results	Advantages	Disadvantages	Considerations and Limitations
conventional methods	Ah OCV EIS	Those that show error values have an RMSE of 1.65% which is a hybrid model with machine learning	No need for an algorithm to implement	Battery needed to be in resting mode for long time	Authors who did not present errors performed simulations or measurement tests. The hybrid algorithm obtained results from physical tests with an embedded algorithm
Methods Based on Filtering Algorithms	KF EKF AKF UKF DKF AUKF	MAE-range of values 0.0426% to 2.22%, RMSE-range of values 0.0044% to 4.58%; these vary depending on the filter used	Acceptable accuracy, dealing with white noise	Need for an accurate enough battery model, extensive time and computational memory, and a complicated algorithm to implement	The method's accuracy is linked to the accuracy of the battery model. Such processes need longer execution time and memory because they have a more complex algorithm

Methods Based on Nonlinear Observers	SMO NLO	MAE-0.928% RMSE-1.7%	Acceptable accuracy and robustness against modeling uncertainties	Difficult for online application due to the complicated computational algorithm	The authors performed physical tests. Still, nonlinear methods are difficult to apply online due to the algorithm's complexity
Methods Based on Learning Algorithms	Technique of machine and deep learning, ANN and their variations, fuzzy logic	MAE-range of values 0.001929% to 3%, RMSE-range of values 0.018% to 3%	Powerful ability to approximate nonlinear functions	Need for a large number of data to train the algorithm applicable for all operating conditions	Applicable for all operating conditions but need a large number of data for algorithm training; they also require more processing time and cost
Hybrid Methods	RLS + AEKF, TNN + I, FUKF, ANN+U KF, ELM+Fuzzy, MHE+KF, ANFIS, SRRLS+EKBF	RMSE-range of values 0.01119% to 3%; these vary depending on the filter used	They reduce the cost of the system but also make the estimation results more effective and reliable	Combining two or three methods is a laborious task and has high complex computation	The combination of the two strategies increases the accuracy of the system

The goal of this work is to compare two different batteries capacities through empirical measurements of battery voltage and current using the Ampere-hour integral estimation method with constant and variable C-rates of discharge current.

II. AMPERE HOUR METHOD

Coulomb Counting method ((known as the current integration method) is chosen and is considered to be the most commonly used to estimate the SOC. It consists of measuring the battery open circuit voltage at start-up to estimate the initial SOC using the battery datasheet information. Then, the battery current is integrated to estimate the amount of charge delivered (or recovered given a charge) by the batteries. This is based on the concept of SOC, as seen in Eq.1, where knowledge of the initial SOC is therefore needed to determine the state of charge [103] [104] [105]. Based on this approach, the SOC for the battery estimate is calculated as follows [106]:

$$SOC = SOC_0 - \frac{1}{C_N} \int_{t_0}^t \eta \cdot I(\tau) d\tau \quad (1)$$

If the initial value of the charge state SOC_0 is specified, or imposed (technically, most researcher imposes the initial state as to be fully charged or fully discharged), the method of coulomb counting becomes very precise and especially simple to determine the SOC [107]. But this method is less accurate if the SOC_0 is unknown.

The coulomb counting strategy computes the remaining stored energy essentially by collecting the charge moved in or out of the battery. The precision of this strategy depends fundamentally from a real estimation of the battery current and precise assessment of the initial SOC. With a pre-knowledge of the initial SOC, which also can be stored at the end of the vehicle trip in a flash memory to be reused as the initial SOC for the next trip (and neglecting the battery self-discharge), the battery SOC can be determined by computing the stored and the released energy flows over the operating time [108].

Nevertheless, the stored energy in the battery is not completely available to be used due to the DOD (Depth of Discharge) which is a quantity of energy to keep inside the battery to avoid its damaging. As a consequence, there are losses during charging and discharging. For more exact SOC assessment, these effect should be considered [109]. Also, the SOC must be recalibrated consistently and discharge limit should be considered for more exact assessment.

III. GENERIC BATTERY MODEL

The battery model can be obtained by considering a voltage source in series with constant resistance, as appeared in Equation.2. [110]

$$V_{batt} = E_0 - K \frac{Q}{Q-i_t} \cdot i_t - R \cdot I + A \cdot \exp(-B \cdot i_t) K \frac{Q}{Q-i_t} \cdot i^* \quad (2)$$

The special feature of this model is the use of filtered current (i^*) through the polarization resistance. This filtered current solves the problem of the algebraic loop due to the simulation of power systems in Simulink.[111].

Although the existence of an hysteresis between the charging and discharging of the battery voltage, this model is still valid for both charge and discharge cycles [112]. The open voltage source is determined with a non-straight condition dependent on the battery genuine SOC.

The battery models can be obtained using:

✓ The charge model ($i^* < 0$)

$$f_1(i_t, i^*, exp, batt\ type) = E_0 - K \frac{Q}{i_t - 0.1Q} \cdot i^* - K \frac{Q}{i_t - Q} \cdot i_t + \exp(t) \quad (3)$$

✓ The discharge model ($i^* > 0$)

$$f_1(i_t, i^*, exp, batt\ type) = E_0 - K \frac{Q}{i_t - Q} \cdot i^* - K \frac{Q}{i_t - Q} \cdot i_t + \exp(t) \quad (4)$$

This model is based on assumptions and has its limitations.

A. Application

✓ 1ST BATTERY

From the discharge curve provided by the manufacturer, the parameters are extracted in Table.2

TABLE II. 1ST BATTERY PARAMETERS

Nominal voltage(V)	12
Rated capacity(Ah)	100
Initial State of charge (%)	100
Maximum capacity(Ah)	110
Fully charged voltage (V)	12.72
Nominal discharge current(A)	10
Internal resistance (ohms)	0.0057
Capacity(Ah)@nominal voltage	59.37
Exponential zone [voltage(V), capacity (Ah)]	[12.48, 10]

✓ 2ND BATTERY

From the discharge curve provided by the manufacturer, the parameters are extracted in Table.3.

TABLE III. 2ND BATTERY PARAMETERS AT C-RATES OF DISCHARGE

C-rates of discharge	0.25C	0.17C	0.09C	different C rates
Nominal Voltage (V)	12	12	12	12
Rated Capacity (Ah)	52	52	52	52
Initial State of Charge (%)	100	100	100	100
Maximum Capacity (Ah)	40.1999	53.8	47.5	47.5
Fully Charged Voltage (V)	12.7	12.8	12.9	12.9
Nominal Discharge Current (A)	13	8.84	4.68	12
Internal Resistance (ohms)	0.0055	0.0055	0.0055	0.0055
Capacity (Ah) @ Nominal Voltage	38.2	44.80	46	46
Exponential Zone (Voltage(V), Capacity (Ah))	[12.7 2.5]	[12.8 1.9]	[12.9 2.225]	[12.9 2.225]

III. EXPERIMENTAL METHOD

A. Battery specification

✓ 1ST BATTERY

The battery used in this work is AGM Lead-Acid battery with nominal voltage of 12V and capacity of 100Ah. This battery's type is chosen because of its availability in our laboratory. While charged and in standby usage, the recommended voltage for individual cell is 2.28 V (6 cells in series). In addition, for cycle usage that requires fast charge, the maximum voltage is 2.45 V. Other characteristics of these batteries are shown in Table.4.

TABLE IV. 1ST BATTERY CHARACTERISTICS

Parameter	Value
Manufacturer	VISION
Model	6FM100E-X
Nominal capacity	100Ah
Nominal voltage	12V
Charging/discharging	Cut-off-voltage(13.8/10.8V)
Recommend charging current (0.25C)	25A

Maximum discharge current	(short time<5s) 900A
Design life	10 years
Operating temperature (charge/discharge)	-10C°-60C°/-20C°-60C°
Shell material	ABS
Weight	29Kg

✓ 2ND BATTERY

The battery used in this research is a valve regulated lead-acid (VRLA) battery with a nominal voltage of 12V and a capacity of 52Ah. The recommended voltage when charged for standby use is 1.75 V for a single cell (6 cells in series). Further features of this battery are shown in Table.5.

TABLE V. 2ND BATTERY CHARACTERISTICS

Cells Per Unit	6
Voltage Per Unit	12 V
Capacity	52 Ah @ 20hr-rate to 1.75V per cell @25°C (77°F)
Weight	Approx. 18 Kg (39.68 lbs.)
Maximum Discharge Current	500A(5sec)
Internal Resistance	Approx. 5.5 mΩ
Operating Temperature Range	Discharge: -15°C~50°C (5°F~122°F) Charge: -15°C~40°C (5°F~104°F) Storage: -15°C~40°C (5°F~104°F)
Float Charging Voltage	13.5 to 13.8 VDC/unit Average at 25°C (77°F)
Recommended Maximum Charging Current Limit	15.6A
Equalization and Cycle Service	14.4 to 15.0 VDC/unit Average at 25°C (77°F)
Self-Discharge	CSB Batteries can be stored for more than 6 months at 25°C (77°F).

B. Measurement test bench

✓ 1ST BATTERY



Fig.2.1st battery Test bench

1. Supply table
2. Lead-acid battery
3. Stabilized power supply
4. MultiMate
5. Current sensor
6. Variable resistor

✓ 2ND BATTERY

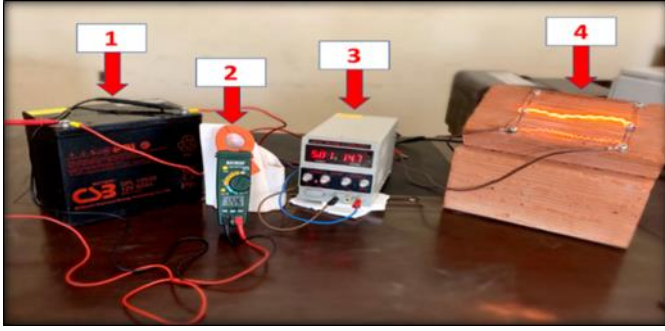


Fig.3. 2nd Battery Test bench

- 1- A Lead Acid Battery
- 2- Clamp Meter
- 3- DC Power Supply
- 4- A Variable Resistor (Handmade)

C. Discharge process test

Two experiments are conducted for this test:

C1. Discharging with constant current (1st battery)

The variable resistor connected with the battery. Previously, the value of the resistor in discharge process is regulated at the value of the current of 10 A. The value of this resistor does not change in discharge process. Furthermore, the voltage and the current are measured every 5 min during the discharge process. After 10 hours and 17 minutes (37020 s), the value of cut-off voltage is achieved to 10.8 V. In this value, the variable resistor is disconnected. These tests were performed under an ambient temperature of 22° C.

C2. Discharging with variable C-rates current (2nd battery)

the variable resistor is connected (to adjust it to a different C-rates discharge current 0.23C, 0.115C then 0.057C respectively. these values of C-rates in discharge current mode correspond to having three resistances connected in parallel. The value of each resistance is 3.225 Ω . During the first period, the discharge is done with 0.23C until 1.11 hours, at this time, the first resistance is disconnected, then the discharge continues with 0.115C until 4.44 hours, at this value the second resistance is disconnected. The discharge is pursued then with 0.057C until

7.23hours, at this time, the last resistance is disconnected and the experiment is stopped. The battery voltage and discharging current are measured each 5 minutes with a clamp meter during 7.23 hours. After 7.23 hours, the end of discharge is reached (cut-off voltage) which is equal to 10.39V. At this point, the discharge resistant is completely disconnected. The ambient temperature equals to $T = 27.1^\circ \text{C}$.

IV. RESULTS AND DISCUSSION

A. 1st Battery

The simulations were performed with the following initial conditions of the discharge: $\text{SOC}_0 = 100\%$, $V_{\text{dis_int}} = 12.6 \text{ V}$, $V_{\text{dis_end}} = 10.8 \text{ V}$ and $T_{\text{dis_sum}} = 36000 \text{ s}$.

Fig.4. shows that the discharge voltage of the battery's datasheet and simulation are slightly different. This difference is due to the imprecision when extracting the parameters of battery model (Table.2) from the curve of the discharge voltage given by the manufacturer. The discharge voltage decreases to 10.8 V for 10 hours which is called the cut-off voltage. The value of 10.8 V does not correspond to the full discharge of battery ($\text{SOC} = 0\%$) but is suitable to minimum SOC value which is equal to nearly 20% [113, 114].

Experimental discharge voltage is fitted to that of the simulation except that the value of the end of discharge voltage (10.8 V) is reached after a period of 10h and 17 minutes (37020 s). This difference indicates that the battery's capacity varies according to the parameters of exploitation (charge mode, the ambient temperature, the discharge current, the battery history ...etc.). Moreover, the voltage given by the manufacturer is related to ideal conditions ($T = 25^\circ \text{C}$, $I_{\text{dis}} = 10 \text{ A}$, the battery State Of Health ($\text{SOH} = 0\%$)). The three curves (datasheet, simulation and experimental) are superposed which indicate that the adopted model of the battery represents the actual behavior of the battery discharge phase with high accuracy.

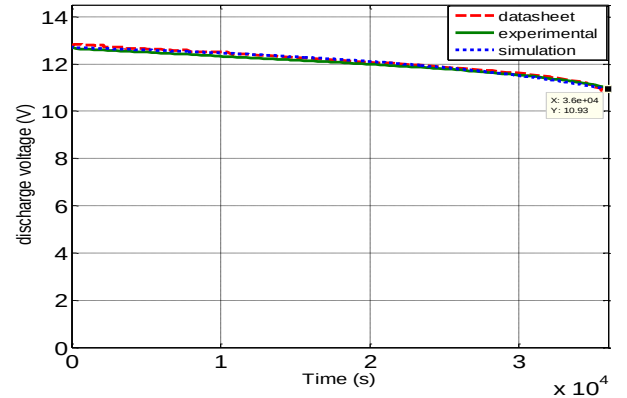


Fig.4. The discharge voltage versus time.

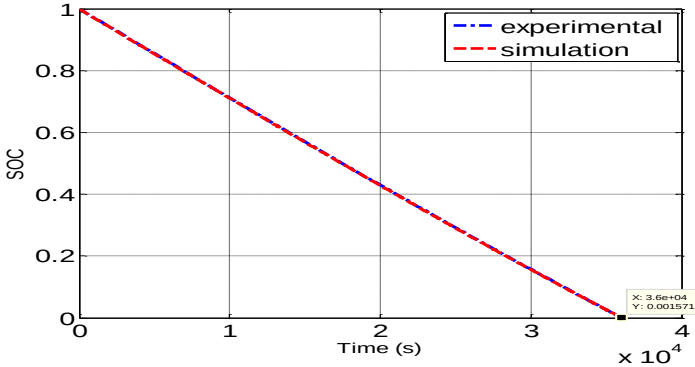


Fig.5. The state of charge (SOC) of the battery versus time

According to Fig.5, we note that the battery is initially full ($SOC_0 = 100\%$) and after 10 hours the state of charge of the battery decreases to the minimum value ($SOC_{min} = 20\%$). This value corresponds to the actual value of the Lead-Acid battery ($SOC_{min} = 20\%$). [113, 115, 114, 108].

Furthermore, this value must not be exceeded in the aim to prevent the damage of the battery. The curve of the SOC estimated by the estimator of Coulomb Counting shows that the discharge of battery is under a constant discharge current ($I_{dis} = 10$ A). Although, this technique follows the state of charge of the battery throughout the discharge time (10 h). SOC_{exp} and SOC_{th} are fitted which indicate that this technique is very precise and robust.

B. 2nd Battery

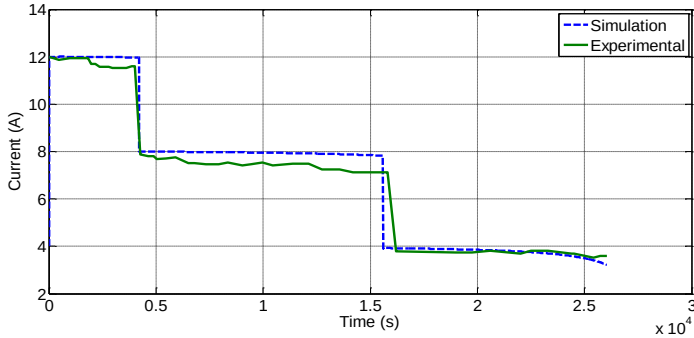


Fig.6. The discharge current as a function of time at different C-rates.

Figure.6 gives the result of the battery discharge using a handmade variable resistor (three equal resistors connected in parallel) where the total equivalent resistor equals to $R_{eq1} = 1.075$ ohms. At the beginning, all resistors are connected and a discharging rate of 0.23C is applied between [0 to 1.11 hours]. Then, one resistance is disconnected from the total equivalent resistors to obtain $R_{eq2} = 2.15$ ohms, corresponding to a discharge rate of 0.115C between [1.11 to 4.44 hours]. Finally, a second resistance is disconnected, which means that only one resistance is kept connected, with a value of 3.225 ohms, corresponding to a discharging rate at 0.057C between [4.44 to 7.23 hours].

Therefore, it is noticed that the current is constant during each C-rates of the discharging range. This is a safe and effective method. Where the minimum value of discharging current is 2.90 A, which corresponds to the value of the cutting voltage of 10.39V.

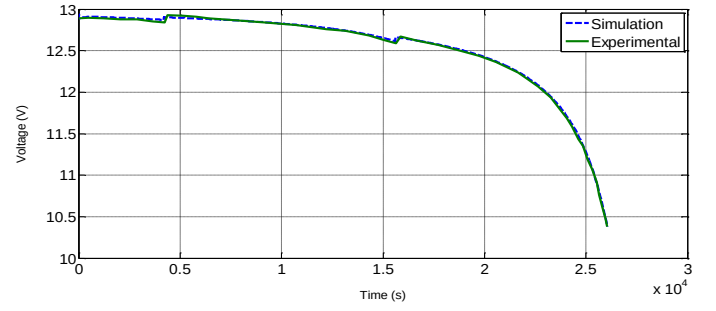


Fig.7. The discharge voltage as a function of time at different C-rates.

Figure.7 illustrates that with three ranges, the discharge voltage decreases to 10.39V for 7.23 hours.

From this curve, it can be noticed that these three ranges express the discharging with different C-rates, 0.23C, 0.115C, and 0.057C respectively.

The two curves (simulated and experimental) are almost the same; this shows that the generic battery presents the real behavior of the battery in the discharging process with a very high precision.

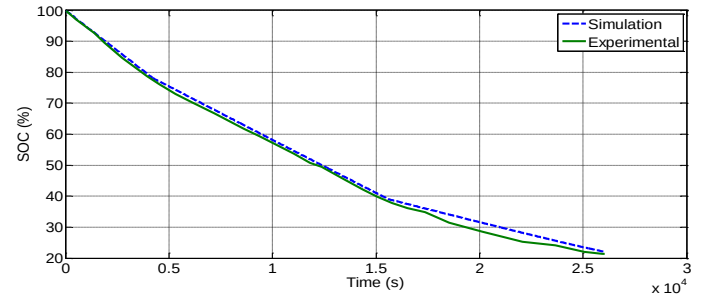


Fig.8. The SOC as a function of time at different C-rates.

The battery is initially fully charged ($SOC_0 = 100\%$) according to figure 8, and the battery charging status decreases after 7.23 hours to the minimum value ($SOC = 20\%$), corresponding to the recommended value of the lead-acid battery ($SOC_{min} = 20\%$), this value must not be exceeded to prevent the battery from being harmed.

The SOC curve calculated by the coulomb counting estimator is a straight line, i.e., the discharge phase is performed under a different discharge current ($I_{disch} = 12A, 6A$ then $3A$, respectively), this technique followed well the state of charge the battery during the entire discharge period (7.23h).

$SOC_{experimental}$ and $SOC_{simulated}$ are almost the same, which shows that this technique is very accurate and usefull.

V. CONCLUSION

Based on the findings, it is evident that:

The discharging process exhibits high accuracy and favorable battery behavior under constant current conditions, particularly crucial for hybrid/electric vehicle applications.

The Ampere hour method proves highly precise in estimating battery state of charge across both constant and variable discharge currents, finding extensive use in embedded applications like hybrid/electric vehicles and renewable energy systems such as photovoltaic setups.

However, drawbacks include inherent inaccuracies and challenges in determining initial state of charge, necessitating the use of precise measurement instruments (voltage, current, temperature sensors) and periodic calibration.

Lead-acid batteries are primarily suited for short-distance vehicles, remaining the most cost-effective battery option per unit of stored energy and likely to maintain

widespread usage in such applications. Many practical and industrial electric vehicles with shorter range requirements can be effectively manufactured.

The proposed method offers simplicity in implementation across portable devices and electric cars, requiring basic calculations and hardware specifications.

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